

# First-principles calculations of the vibrational spectra of one-dimensional C<sub>60</sub> polymers

Titus A. Beu\*

Faculty of Physics, University "Babeş-Bolyai," 3400 Cluj-Napoca, Romania

Jun Onoe

Research Laboratory for Nuclear Reactors, Tokyo Institute of Technology, 2-12-1 O-okayama, Tokyo 152-8550

(Received 24 August 2006; revised manuscript received 4 October 2006; published 20 November 2006)

The vibrational properties of several C<sub>60</sub> dimers belonging to the Stone-Wales rearrangement sequence [described by E. Osawa and K. Honda, Full. Sci. Technol. **4**, 939 (1996)], along with a trimer and a tetramer isomer are investigated by tight-binding and density functional calculations. The IR absorption bands found by Onoe, Nakayama, Aono and Hara [Appl. Phys. Lett. **82**, 595 (2003)] in the spectra of electron-beam irradiated C<sub>60</sub> films are explained and attributed mainly to surface vibrations located in the waist region of the considered polymers. The features of the Raman spectra are also discussed and related to experimental spectra of photo-polymerized C<sub>60</sub>.

DOI: 10.1103/PhysRevB.74.195426

PACS number(s): 71.15.Mb, 71.20.Tx

## I. INTRODUCTION

The polymerization of C<sub>60</sub> induced by electron-beam (EB) irradiation has proven to be a viable technique for nanofabrication of C<sub>60</sub>-based carbon materials, potentially useful for developing carbon-based nanodevices. Onoe and others have extensively investigated EB polymerized C<sub>60</sub> films by *in situ* Fourier-transform infrared spectroscopy, *in situ* x-ray photoelectron spectroscopy, and *in situ* scanning tunneling microscopy.<sup>1-5</sup> In EB polymerized films, C<sub>60</sub> molecules coalesce to form tubular linkages and the metallic-type resistivity, found by *ex situ* four-probe measurements to be as low as 7 Ω cm, is considerably smaller than that of the insulating solid C<sub>60</sub> (10<sup>8</sup>–10<sup>14</sup> Ω cm).<sup>3</sup>

The measured IR spectra show that as polymerization proceeds, supplementary peaks develop around 1340, 1360, and 1390 cm<sup>-1</sup> as compared to the pristine C<sub>60</sub> film.<sup>2-4</sup> Vibrational calculations based on the AM1 semiempirical method have been carried out by Hara and Onoe for several peanut-shaped C<sub>120</sub> isomers.<sup>4</sup> Qualitatively, the results have evidenced that the IR absorption band observed in the range 1300–1400 cm<sup>-1</sup> can be attributed to the normal modes involving mainly atomic in-plane motions located in the waist region. Earlier, very interesting density-functional tight-binding calculations of vibrational properties of C<sub>60</sub> oligomers have been published by Porezag and Fravenheim<sup>6</sup>

The present investigations are primarily intended to explain in more detail the observed IR spectra of EB irradiated C<sub>60</sub> films by means of calculations of vibrational properties of one-dimensional C<sub>60</sub> polymers. The described calculations are a sequel to the previously published work concerning the electronic structure of one-dimensional C<sub>60</sub> polymers,<sup>7</sup> aimed at explaining the metallic behavior of EB irradiated films and providing the structural data. Both the present and the previous research rely on two complementary approaches: (a) efficient tight-binding (TB) parametrizations and (b) all-electron density functional theory (DFT) with carefully assessed exchange-correlation functionals.

Since the computational effort required to obtain IR and, in particular, Raman spectra for C<sub>120</sub> peanuts lacking any

exploitable symmetry properties is a real challenge even for up-to-date machines, our vibrational DFT calculations are limited to the only isomer of the Stone-Wales rearrangement path (besides the capped nanotube), which shows any symmetry at all, i.e., the C<sub>2h</sub>-symmetry peanut designated in our previous paper as P4. Not being subject to such constraints, the TB calculations are feasible in reasonable amounts of time for larger polymers, as such we have investigated several of the Stone-Wales C<sub>120</sub> isomers, along with the trimer (C<sub>180</sub>) and tetramer (C<sub>240</sub>) derived from P4. The vibrational TB and DFT results turn out to be consistent, offering a truthful explanation of the observed vibrational spectra.

## II. CALCULATION METHODS

We have successfully employed the TB method over the last years for describing structural and vibrational properties of the C<sub>36</sub>, C<sub>60</sub>, and C<sub>70</sub> fullerenes.<sup>7-9</sup> The TB approaches can be viewed as simplified two-center-oriented *ab initio* methods, with the electronic properties calculated from a parametrized representation of the Kohn-Sham equation. The TB parametrization of Papaconstantopoulos *et al.*<sup>10</sup> describes the environment of each atom by a pseudo-atomic density,

$$\rho_I = \sum_{J \neq I}^N \exp(-\lambda^2 R_{IJ}) f(R_{IJ}),$$

which depends exponentially on the distances to all neighbors and features a cutoff function

TABLE I. HOMO-LUMO gap  $\Delta$  and (longitudinal) extents  $L$  of C<sub>60</sub> and its dimer P4 obtained from DFT calculations with the basis sets 6–31 g(d) and 6–31 g, respectively. The differences between the basis sets are indicated in parentheses.

	Basis set	$\Delta$ (eV)	$L$ (Å)
C <sub>60</sub>	6–31 g(d)	1.668	6.987
	6–31 g	1.770 (0.10)	7.013 (0.03)
C <sub>120</sub> (P4)	6–31 g(d)	0.333	15.99
	6–31 g	0.379 (0.05)	16.04 (0.05)

TABLE II. Frequencies and intensities (activities) of the IR and Raman active vibrational modes of  $C_{60}$  resulted from DFT calculations and experiment.

	6-31 g(d)		6-31 g		Exp.	
$T_{1u}$	513	22.3	527	21.2	527 <sup>a</sup>	1.00
$T_{1u}$	577	11.9	575	10.2	576 <sup>a</sup>	0.30
$T_{1u}$	1193	8.8	1186	9.91	1183 <sup>a</sup>	0.22
$T_{1u}$	1447	9.2	1464	14.2	1429 <sup>a</sup>	0.24
$A_g$	488	143	487	138	493 <sup>b</sup>	0.02
$A_g$	1487	312	1506	365	1468 <sup>b</sup>	0.10
$H_g$	257	24.2	262	23.9	270 <sup>b</sup>	0.52
$H_g$	419	0.27	430	0.43	430 <sup>b</sup>	0.40
$H_g$	686	0.27	700	0.44	708 <sup>b</sup>	0.40
$H_g$	771	8.68	771	7.55	772 <sup>b</sup>	0.38
$H_g$	1101	11.4	1103	11.6	1099 <sup>b</sup>	0.01
$H_g$	1255	18.1	1250	20.6	1248 <sup>b</sup>	0.01
$H_g$	1435	2.35	1445	2.95	1426 <sup>b</sup>	0.44
$H_g$	1570	47.1	1581	52.0	1573 <sup>b</sup>	0.52

<sup>a</sup>Reference 19.

<sup>b</sup>Reference 20.

$$f(R) = \frac{1}{1 + \exp[(R - R_c)/\Delta]}.$$

The parameters employed for carbon are  $R_c = 10.5 a_0$  and  $\Delta = 0.5 a_0$  ( $a_0$  is the Bohr radius).

The (diagonal) *on-site* Hamiltonian elements are parametrized in terms of the local pseudoatomic density by

$$h_l^i = \alpha_l + \beta_l \rho_l^{2/3} + \gamma_l \rho_l^{4/3} + \chi_l \rho_l^2,$$

where  $l=s,p$ . The two-center Slater-Koster hopping terms are defined as polynomials with an exponential cutoff ( $\mu = \sigma, \pi$ )

$$H_{ll',\mu}(R) = (a_{ll',\mu} + b_{ll',\mu}R + c_{ll',\mu}R^2) \exp(-d_{ll',\mu}^2 R) f(R)$$

$$S_{ll',\mu}(R) = (\delta_{ll'} + p_{ll',\mu}R + q_{ll',\mu}R^2 + r_{ll',\mu}R^3) \times \exp(-s_{ll',\mu}^2 R) f(R).$$

The approach we used to analyze the vibrational properties within the harmonic approximation is based on the dynamical matrix.<sup>11</sup> For this approach to be applicable, one has to start from a fully relaxed geometrical structure of the molecular aggregate under consideration. The elements of the dynamical matrix have the significance of local curvatures of the potential energy surface at the atom sites, and they can be approximated by

$$\mathcal{H}_{IJ}^{\alpha\beta} = -\frac{F_I^\alpha(r_J^\beta + h) - F_I^\alpha(r_J^\beta - h)}{2h} + O(h^2),$$

$$I, J = 1, 2, \dots, N, \quad \alpha, \beta = x, y, z,$$

$F_I^\alpha(r_J^\beta \pm h)$  being the  $\alpha$  component of the force acting on atom  $I$  when atom  $J$  is displaced by  $\pm h$  along direction  $\beta$ .

In order to obtain the vibrational frequencies and the corresponding atomic displacements for the normal modes, one has to solve then the generalized eigenvalue problem for the dynamical matrix

$$\mathcal{H} \cdot \delta \mathbf{r}_k = \omega_k^2 \mathbf{M} \cdot \delta \mathbf{r}_k, \quad k = 1, 2, \dots, 3N.$$

Here,  $\omega_k$  is the eigenvalue (frequency) of normal mode  $k$ ,  $\delta \mathbf{r}_k$  is the corresponding eigenvector (having as components the atomic displacements), and  $\mathbf{M}$  is the diagonal matrix with the atomic masses on the main diagonal.

Our DFT calculations have been carried out using the package GAUSSIAN 03.<sup>12</sup> In our previous paper,<sup>7</sup> we have provided evidence for the superiority of two recent functionals in calculations regarding fullerenes as compared to the traditional B3LYP functional.<sup>13</sup> The employed PBE exchange and, respectively, gradient-corrected correlation func-

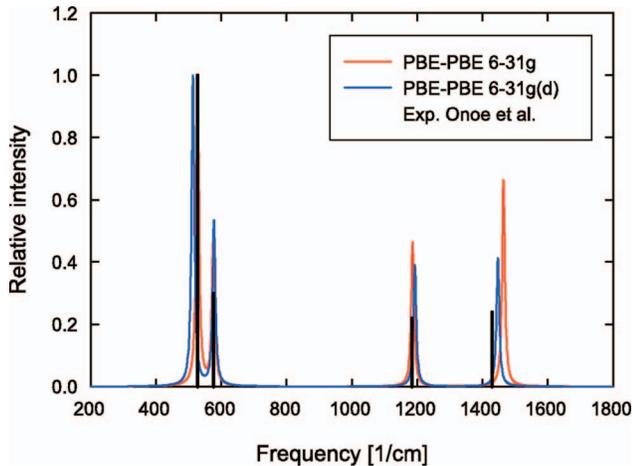


FIG. 1. (Color) IR-spectrum of the  $C_{60}$  fullerene resulted from DFT calculations with the PBE-PBE exchange-correlation functional and the 6-31 g and 6-31 g(d) basis sets, along with the experimental results of Onoe *et al.*<sup>3,5</sup> (drop lines).

TABLE III. TB vibrational results for several  $C_{60}$  peanuts belonging to the Stone-Wales rearrangement pathway, and for the trimer and tetramer derived from the P4 peanut. Only the three most intense lines between 1300 and 1400  $\text{cm}^{-1}$  are included for each species.  $\Delta$  designates the HOMO-LUMO gap and  $I$  is the IR intensity.

$\Delta$ (eV)	$\nu$	$I$	sym.	$\Delta$ (eV)	$\nu$	$I$	sym.	$\Delta$ (eV)	$\nu$	$I$	sym.
	P4				P7				P8		
0.294	1317	0.05	$A_u$	0.587	1317	0.06	A	0.370	1300	0.17	A
	1333	0.06	$E_u$		1338	0.08	A		1333	0.06	A
	1381	0.16	$E_u$		1375	0.11	A		1379	0.13	A
	P12				P16				P20		
0.273	1316	0.20	A	0.256	1302	0.05	A	0.184	1302	0.04	A
	1340	0.10	A		1322	0.05	A		1315	0.04	A
	1389	0.05	E		1338	0.06	A		1357	0.03	A
	P24				$C_{180}$				$C_{240}$		
0.566	1314	0.06	$E_u$	0.050	1317	0.03	$A_u$	0.195	1363	0.05	$A_u$
	1379	0.17	$E_u$		1382	0.17	$A_u$		1380	0.12	$E_u$
	1399	0.17	$E_u$		1392	0.01	$E_u$		1389	0.02	$A_u$

tionals of Perdew *et al.*,<sup>14</sup> and the HCTH exchange functional including gradient-corrected correlation of Hamprecht *et al.*<sup>15</sup> reproduce fairly the experimental HOMO-LUMO gap of  $C_{60}$  (1.6–1.85 eV),<sup>16</sup> i.e., 1.668 eV (PBE), and 1.698 eV (HCTH). In the present work, we have used only the PBE functional, the corresponding results yielded by the HCTH functional being insignificantly different.

Since all-electron DFT calculations of the IR and Raman spectra of  $C_{60}$  dimers employing the split valence polarized basis set 6–31 g(d) (used throughout in our previous paper) have turned out to involve excessive computer resources, we have performed all production calculations reported here using the subset 6–31 g. A comparison of the accuracy of the structural results provided by the two basis sets is given in Table I, with regard to the HOMO-LUMO gap and the geometrical extents of  $C_{60}$  and its dimer P4. Although the relative differences between the 6–31 g and the 6–31 g(d) results reach as much as 12.1% for the HOMO-LUMO gap in

the case of the P4 dimer (0.05 eV), the geometrical extents coincide within <0.4% (0.05 Å).

In order to assess the way the two basis sets perform in spectroscopy applications, we have carried out also preliminary calculations regarding the vibrational modes of  $C_{60}$ . The comparison of the calculated and experimental frequencies (Table II) emphasizes a rather reassuring aspect: as compared to the 6–31 g(d) basis set, the 6–31 g subset seems to reproduce equally well three of the four IR-active modes ( $T_{1u}$  symmetry) and five of the ten Raman-active modes ( $A_g$  and  $H_g$  symmetry).

It is a matter of common knowledge that first-principles methods typically tend to overestimate the vibrational frequencies (by up to 10%), mainly due to inherent approximations in the electron correlation, and DFT methods are no exception to this behavior.<sup>17</sup> It should be noted, however, that we employed no scaling and that all frequencies of  $C_{60}$  were reproduced within <3%. For the lowest three IR active modes, the absolute deviations do not exceed 4  $\text{cm}^{-1}$ .

The reasonable deviations with respect to the experiment caused by the smaller 6–31 g basis set, obtained nevertheless at a significantly lower computational effort, can be judged also visually from Fig. 1, where the measurements of Hara *et al.*,<sup>1</sup> represented as drop lines, are used as reference.

### III. RESULTS AND DISCUSSION

Seven of the nine  $C_{60}$  polymers investigated in this study are dimers belonging to the Stone-Wales rearrangement sequence, described by Osawa and Honda<sup>18</sup> and characterized by the fact that the interconversion between successive isomers can be achieved merely by rotations of atom pairs about their bond center. The dimers considered here are namely P4, P7, P8, P12, P16, P20, and the capped nanotube P24. Only four of all polymers show well-defined symmetry: the P4 peanut, along with the trimer ( $C_{180}$ ) and the tetramer ( $C_{240}$ ) derived from it have  $C_{2h}$  symmetry, while the P24

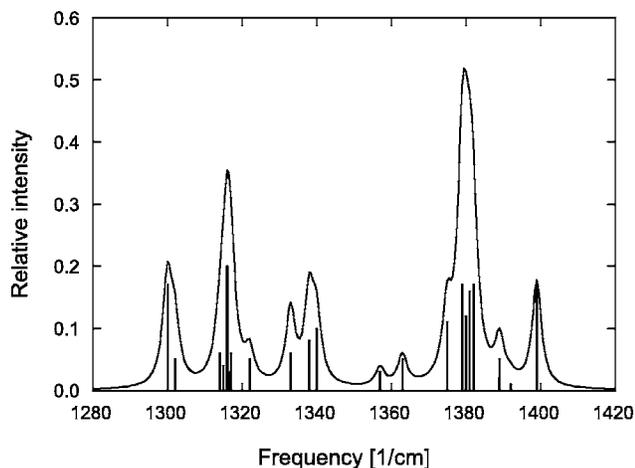


FIG. 2. IR-lines for the peanuts P4, P7, P8, P12, P16, P20, and P24 and convoluted spectrum of corresponding Lorentzian peaks resulted from TB calculations.

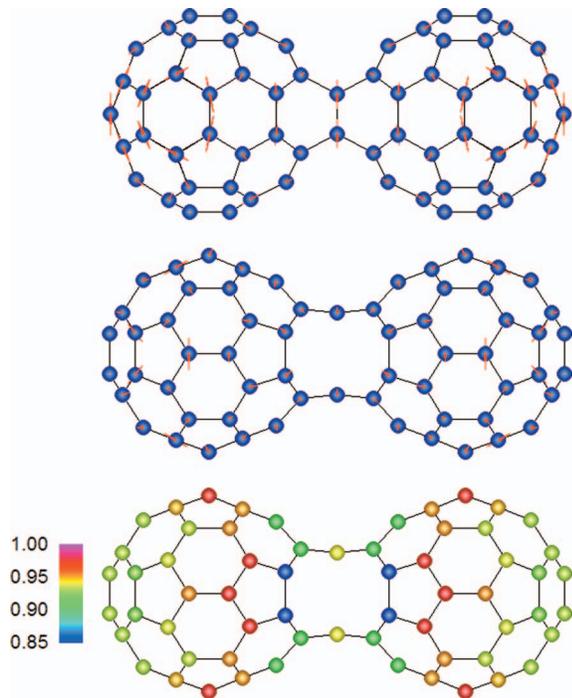


FIG. 3. (Color) Atomic displacements for the IR-active vibrational mode at  $1333\text{ cm}^{-1}$  and pseudoatomic density for the P4 peanut yielded by TB calculations.

nanotube shows  $C_5$  symmetry. The geometrical structure of all polymers was optimized corresponding to the tight-binding force field by simulated annealing embedded in molecular dynamics.

Table III summarizes all relevant TB results for the considered polymers. Besides the HOMO-LUMO gap, the frequency, intensity and symmetry of the three most intense IR-active modes located in the interval  $1300\text{--}1400\text{ cm}^{-1}$  (where the major changes in the electron-beam irradiated  $C_{60}$  films were observed) are listed for each isomer. Because of their symmetry properties, the P4 and the P24 peanuts, together with the trimer and tetramer derived from P4, feature *ungerade* IR-active vibrations. As general findings, all struc-

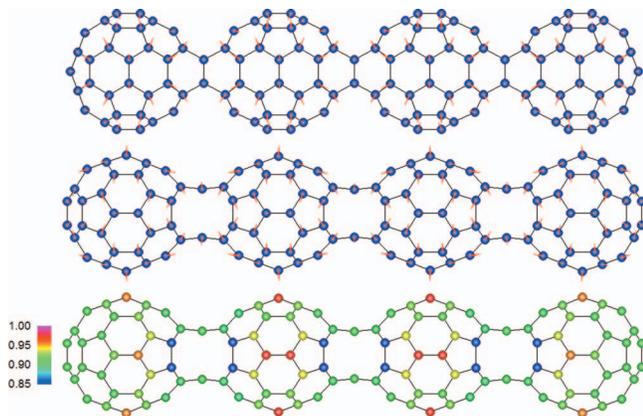


FIG. 4. (Color) Atomic displacements for the IR-active vibrational mode at  $1380\text{ cm}^{-1}$  and pseudoatomic density for the tetramer derived from the P4 peanut, yielded by TB calculations.

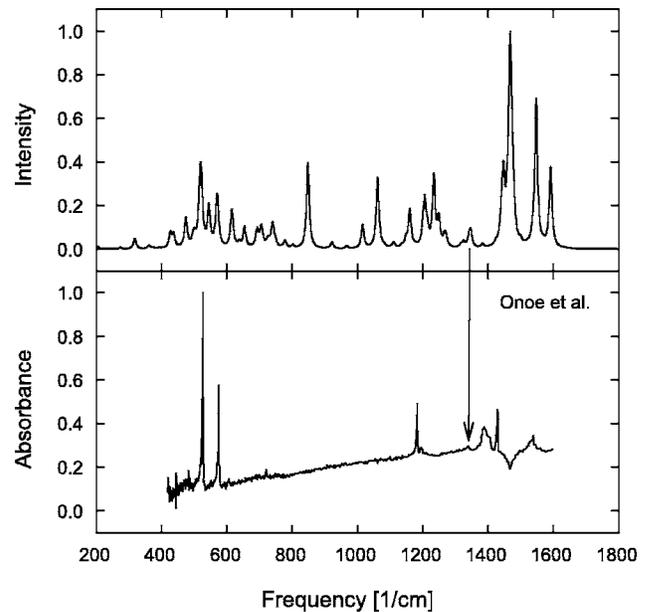


FIG. 5. IR spectrum of the P4 peanut obtained from DFT calculations with the PBE-PBE exchange-correlation functional and the 6–31 g basis set, along with the experimental spectrum of Onoe *et al.*<sup>1</sup>

tures investigated possess at TB level IR lines in the interval  $1300\text{--}1400\text{ cm}^{-1}$  and the most intense vibrations are predominantly surface modes.

Even though there are no obvious correlations between the position of the peanuts in the Stone-Wales sequence and the frequency, intensity, or symmetry of their vibrations, Fig. 2, where the convoluted spectrum of all investigated polymers is depicted, clearly shows that more intense bands are formed around  $1320$ ,  $1340$ , and  $1380\text{ cm}^{-1}$ . The convoluted spectrum was obtained by assigning Lorentzian peaks of half width  $1\text{ cm}^{-1}$  to the calculated TB frequencies and, by a small rescaling of  $0.98$ , this spectrum provides a consistent explanation of the experimental spectrum. For the bands around  $1320$  and  $1340\text{ cm}^{-1}$ , the P12 peanut, located just in the middle of the Stone-Wales rearrangement pathway, seems to be mainly responsible. The experimental absorption

TABLE IV. IR active vibrational modes of the peanut P4 in the interval  $1300\text{--}1400\text{ cm}^{-1}$  obtained with the PBE-PBE exchange-correlation functional and the 6–31 g basis set.

$\nu$	IR intensity	symmetry
1307.08	0.14	$A_u$
1316.11	0.83	$B_u$
1316.67	1.05	$B_u$
1324.68	3.62	$A_u$
1331.55	0.40	$B_u$
1341.56	2.79	$B_u$
1342.26	4.31	$B_u$
1347.16	9.43	$A_u$
1382.83	2.39	$B_u$

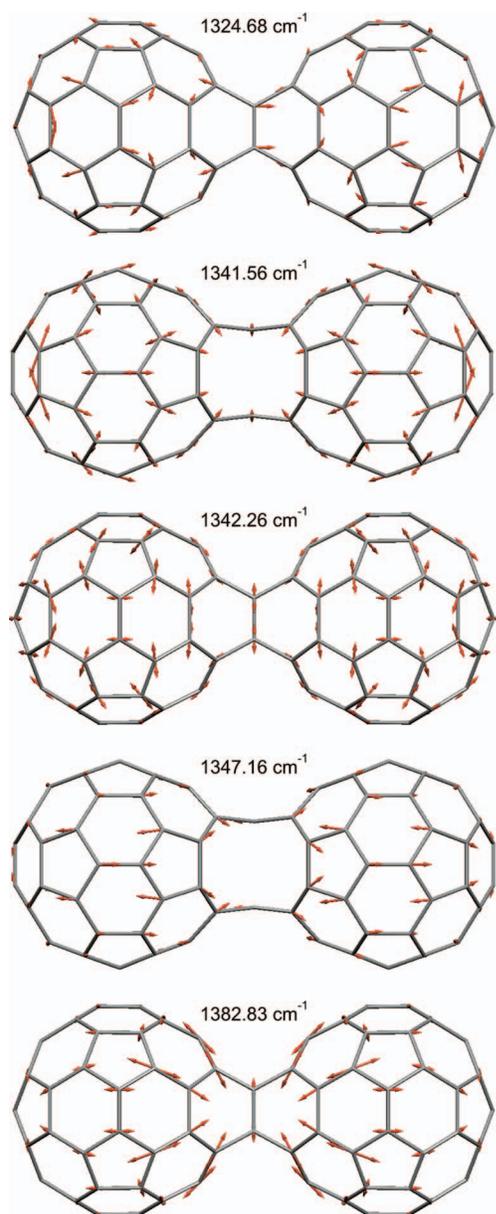


FIG. 6. (Color) Most intense IR-active vibrational modes of the P4 peanut between 1300 and 1400  $\text{cm}^{-1}$  yielded by DFT calculations with the exchange-correlation functional PBE-PBE and the 6-31 g basis set.

maximum ( $\sim 1380 \text{ cm}^{-1}$ ) can be well explained especially by the small peanuts (up to P8 and including the polymers derived from the symmetric peanut P4).

Since visualizing normal vibrations of molecular structures without symmetry properties is inherently difficult, we have chosen to illustrate the TB results by normal modes of the P4 peanut (Fig. 3) and of the tetramer derived from P4 (Fig. 4). Both normal modes are located in the interval of interest (between 1300 and 1400  $\text{cm}^{-1}$ ). The vibration of the P4 peanut located at 1333  $\text{cm}^{-1}$  is seen to be predominantly a surface mode involving mainly the longitudinal belt of hexagons comprising the waist region (upper view). The pseudoatomic density reaches its minimum (blue) at atomic sites located in the waist region. The normal mode of the

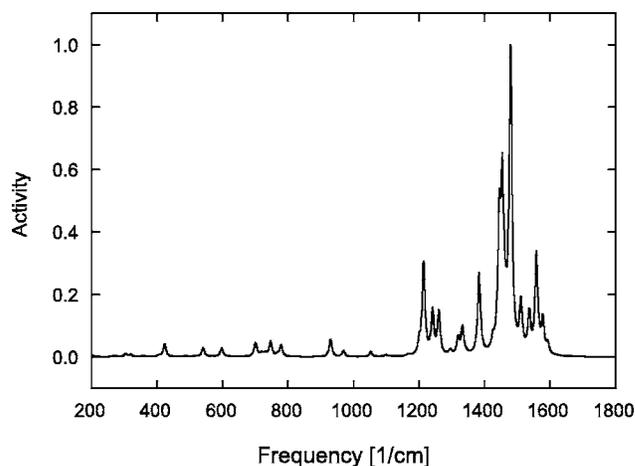


FIG. 7. Raman spectrum of the P4 fullerene obtained from DFT calculations with the exchange-correlation functional PBE-PBE and the basis set 6-31 g.

tetramer derived from P4 is also largely a surface mode, with low pseudoatomic density in the waist regions.

As already pointed out, our DFT calculations have been carried out employing the PBE-PBE exchange-correlation functional of Perdew *et al.*<sup>14</sup> and, for feasibility reasons, the 6-31 g basis set, which has proven to provide vibrational results of comparable quality with the larger 6-31 g(d) basis set. The calculations have been limited to the P4 peanut, whose symmetry properties can be exploited in reducing significantly the computation time.

Figure 5 shows the calculated IR spectrum (without any scaling), along with the experimental spectrum of an electron-beam irradiated  $\text{C}_{60}$  film on a CsI substrate published by Onoe *et al.*<sup>1</sup> It is easily noticeable that the experimental absorption peak developing around 1340  $\text{cm}^{-1}$  as a consequence of the electron-beam irradiation has a matching peak in the calculated spectrum (the two peaks are connected by an arrow). Besides the fair agreement of the DFT and TB results concerning the existence of a peak around 1340  $\text{cm}^{-1}$ , the symmetric P4 peanut alone offers already an explanation of the IR behavior of the coalesced structures formed by electron-beam irradiation.

As can be seen from Table IV, where the normal modes with the frequency comprised between 1300 and 1400  $\text{cm}^{-1}$  have been listed, the most intense IR-active mode is located at 1347  $\text{cm}^{-1}$ , having  $A_u$  symmetry and more than the double intensity as compared to the rest of the normal modes. Moreover, having in view only the five most intense modes, it is

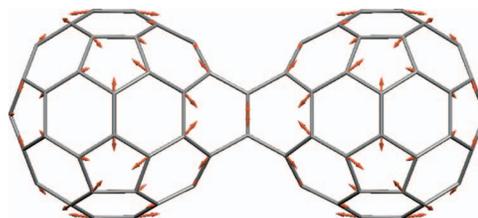


FIG. 8. (Color) Raman-active vibrational mode of the P4 peanut at 1510.76  $\text{cm}^{-1}$  corresponding to the  $A_g(2)$  pinch mode of the  $\text{C}_{60}$  fullerene located at 1469  $\text{cm}^{-1}$ .

interesting to note that the DFT calculations on the P4 peanut confirm the appearance of three groups of IR-active modes (already suggested by the TB results), located, respectively, around 1320, 1340, and 1380  $\text{cm}^{-1}$ .

Figure 6 shows the displacement patterns of the five most intense IR-active modes, and these appear to be predominantly surface modes. In particular, the most intense mode at 1347  $\text{cm}^{-1}$  is predominantly a *longitudinal* surface mode. The surface character of the most intense modes is again in agreement with the TB findings, holding also for the rest of the considered polymers.

Even though no Raman spectra for electron-beam irradiated  $\text{C}_{60}$  films are available in the literature, we have depicted in Fig. 7 the Raman spectrum resulted for the P4 peanut by our DFT calculations. The most significant Raman-activity occurs between 1200 and 1600  $\text{cm}^{-1}$ . The most pronounced peak appears at 1479  $\text{cm}^{-1}$ , in the range of the  $A_g(2)$  pinch mode of pristine  $\text{C}_{60}$  (located at 1469  $\text{cm}^{-1}$ ). However, the normal mode corresponding to the  $\text{C}_{60}$  pinch mode is actually the one located at 1510  $\text{cm}^{-1}$  and its displacement pattern can be seen in Fig. 8. Applying the scaling factor 0.975 suggested by the calculations for  $\text{C}_{60}$  (see Table II), the “pinch” mode of the P4 peanut repositions at 1473  $\text{cm}^{-1}$ , which overestimates by  $<0.7\%$  the measurements of Burger *et al.*<sup>21</sup> on photopolymerized  $\text{C}_{60}$ .

It should be noted that the scaling applied to our Raman spectrum is comprised within the range commonly consid-

ered to be moderate (0.95–1), Raman spectra being known to be particularly sensitive to the employed combination of basis set and exchange-correlation functional.<sup>17</sup>

As suggested by the results for  $\text{C}_{60}$ , the basis set 6–31 g(d) would perform better in the particular case of the Raman spectrum.

#### IV. CONCLUSIONS

The vibrational properties of several one-dimensional  $\text{C}_{60}$  polymers have been investigated by tight-binding and density functional theory calculations. Both approaches provide a consistent explanation of the measured IR-spectra of electron-beam irradiated  $\text{C}_{60}$  films, mainly, the existence of a significant absorption maximum around 1340  $\text{cm}^{-1}$ . Additional peaks are predicted by both approaches around 1320 and 1380  $\text{cm}^{-1}$ . The most intense IR-active vibrational modes between 1300 and 1400  $\text{cm}^{-1}$  have been found to be predominantly surface modes. By slight scaling, the DFT calculations reproduce the features developing in the Raman spectra of photopolymerized  $\text{C}_{60}$ .

#### ACKNOWLEDGMENT

One of the authors (T.A.B.) gratefully acknowledges a visiting professor position at the Tokyo Institute of Technology.

\*Electronic address: tbeu@phys.ubbcluj.ro

- <sup>1</sup>T. Hara, J. Onoe, H. Tanaka, Y. Li, and K. Takeuchi, *Jpn. J. Appl. Phys., Part 1* **39**, 1872 (2000).
- <sup>2</sup>T. Hara, J. Onoe, and K. Takeuchi, *J. Appl. Phys.* **92**, 7302 (2002).
- <sup>3</sup>J. Onoe, T. Nakayama, M. Aono, and T. Hara, *Appl. Phys. Lett.* **82**, 595 (2003).
- <sup>4</sup>T. Hara and J. Onoe, *Eur. Phys. J. D* **24**, 389 (2003).
- <sup>5</sup>J. Onoe, T. Nakayama, M. Aono, and T. Hara, *J. Phys. Chem. Solids* **65**, 343 (2004).
- <sup>6</sup>D. Porezag and T. Frauenheim, *Carbon* **37**, 463 (1999).
- <sup>7</sup>T. A. Beu, J. Onoe, and A. Hida, *Phys. Rev. B* **72**, 155416 (2005).
- <sup>8</sup>T. A. Beu, J. Onoe, and K. Takeuchi, *Eur. Phys. J. D* **10**, 391 (2000).
- <sup>9</sup>T. A. Beu, J. Onoe, and K. Takeuchi, *Eur. Phys. J. D* **17**, 205 (2001).
- <sup>10</sup>D. A. Papaconstantopoulos, M. J. Mehl, S. C. Erwin, and M. R. Pederson, in *Tight-Binding Approach to Computational Materials Science*, edited by P. Turchi, A. Gonis, and L. Colombo, MRS Symposia Proceedings No. 491 (Materials Research Society, Warrendale, PA, 1998), p. 221.
- <sup>11</sup>Th. Köhler, Th. Frauenheim, and G. Jungnickel, *Phys. Rev. B* **52**, 11837 (1995).
- <sup>12</sup>M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, J. A. Montgomery, Jr., T. Vreven, K. N. Kudin, J. C. Burant, J. M. Millam, S. S. Iyengar, J. Tomasi, V. Barone, B. Mennucci, M. Cossi, G. Scalmani, N. Rega, G. A. Petersson, H. Nakatsuji, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, M. Klene, X. Li, J. E. Knox, H. P. Hratchian, J.

- B. Cross, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, P. Y. Ayala, K. Morokuma, G. A. Voth, P. Salvador, J. J. Dannenberg, V. G. Zakrzewski, S. Dapprich, A. D. Daniels, M. C. Strain, O. Farkas, D. K. Malick, A. D. Rabuck, K. Raghavachari, J. B. Foresman, J. V. Ortiz, Q. Cui, A. G. Baboul, S. Clifford, J. Cioslowski, B. B. Stefanov, G. Liu, A. Liashenko, P. Piskorz, I. Komaromi, R. L. Martin, D. J. Fox, T. Keith, M. A. Al-Laham, C. Y. Peng, A. Nanayakkara, M. Challacombe, P. M. W. Gill, B. Johnson, W. Chen, M. W. Wong, C. Gonzalez, and J. A. Pople, *GAUSSIAN 03*, REVISION B.05 (Gaussian, Inc., Pittsburgh PA, 2003).
- <sup>13</sup>A. D. Becke, *J. Chem. Phys.* **98**, 5648 (1993); C. Lee, W. Yang, and R. G. Parr, *Phys. Rev. B* **37**, 785 (1988).
- <sup>14</sup>J. P. Perdew, K. Burke, and M. Ernzerhof, *Phys. Rev. Lett.* **77**, 3865 (1996).
- <sup>15</sup>F. A. Hamprecht, A. J. Cohen, D. J. Tozer, and N. C. Handy, *J. Chem. Phys.* **109**, 6264 (1998).
- <sup>16</sup>R. K. Kremer, *Appl. Phys. A: Solids Surf.* **56**, 211 (1993).
- <sup>17</sup>J. B. Foresman and AEleen Fischer, *Exploring Chemistry With Electronic Structure Methods: A Guide to Using Gaussian*, 2nd Ed. (Gaussian, Inc., Pittsburgh, 1996), pp. 63–64.
- <sup>18</sup>E. Osawa and K. Honda, *Fullerene Sci. Technol.* **4**, 939 (1996).
- <sup>19</sup>K. Esfarjani, Y. Hashi, J. Onoe, K. Takeuchi, and Y. Kawazoe, *Phys. Rev. B* **57**, 223 (1998).
- <sup>20</sup>K. A. Wang, Y. Wang, P. Zhou, J. M. Holden, S. L. Ren, G. T. Hager, H. F. Ni, P. C. Eklund, G. Dresselhaus, and M. S. Dresselhaus, *Phys. Rev. B* **45**, 1955 (1992).
- <sup>21</sup>B. Burger, J. Winter, and H. Kuzmany, *Z. Phys. B: Condens. Matter* **101**, 227 (1996).