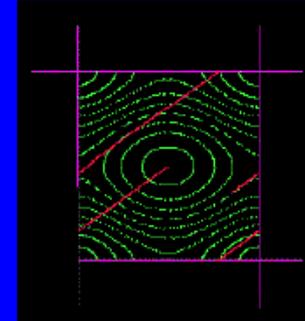
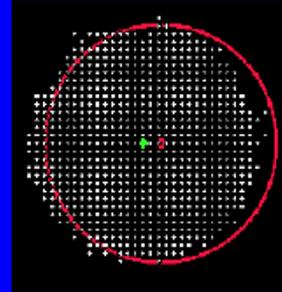
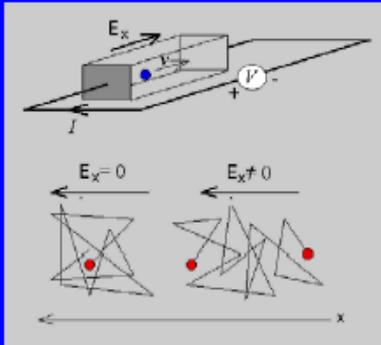


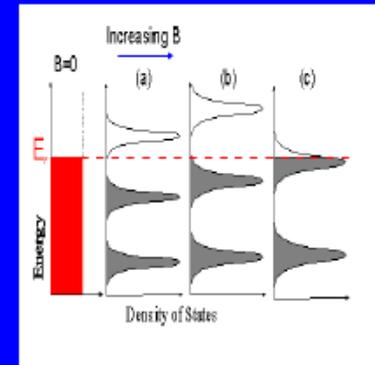
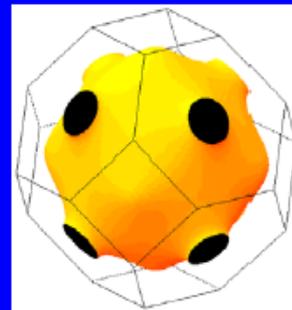
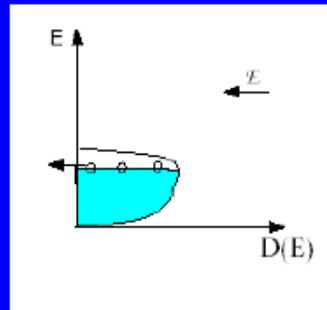
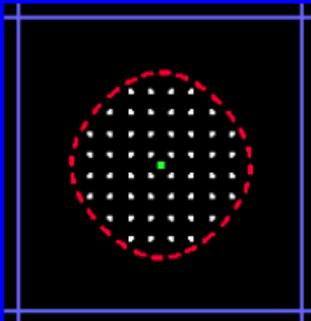
Transport Phenomena in Solids

Motions of electrons and transport phenomena



$$\sigma = \frac{ne^2\tau}{m}$$

$$\left(\frac{1}{m^*}\right)_{ij} = \frac{1}{\hbar^2} \sum_j \frac{\partial^2 E(\vec{k})}{\partial k_i \partial k_j}$$



the study of transport phenomena in physics is related with the exchange of mass, energy, and momentum studied systems.

- fluid mechanics, heat transfer, and mass transfer
- in solid state physics, the motion and interaction of electrons, holes and phonons are studied under "transport phenomena".

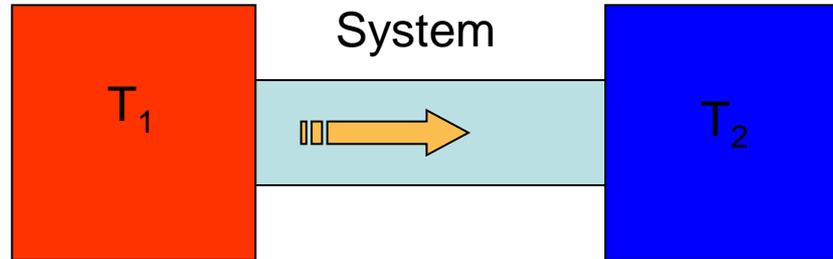
Content:

1. Introduction. General. Fick' Law. Boltzmann Eq. Relaxation time.
2. Electronic transport in conductors. Electron-phonon scattering.
3. Electron-imperfection scattering
4. Electrical conductivity. Bloch-Gruneisen
5. Magnetic scattering
6. Thermal conductivity
7. Thermoelectric phenomena
8. Electrical conductivity in magnetic fields.
9. Anomalous Hall effect
10. Magnetoresistance : AMR, CMR
11. Magnetoresistance : GMR TMR.
12. Strongly correlated electron systems

1. C. Kittel, H. Kroemer, *Thermal Physics*, W.H. Freeman Co. New York
2. C. Kittel, *Introduction to Solid State Physics* (7-8 ed., Wiley, 1996)
3. N. W. Ashcroft, N. D. Mermin, *Solid State Physics*, 1976.
4. U. Mizutani, *Introduction to the Electron Theory of Metals*, Cambridge University Press 2001.
5. Ch. Enss, S. Hunklinger, *Low-Temperature Physics*, Springer-Verlag Berlin Heidelberg 2005.
6. M. Coldea, *Magnetorezistentia. Efecte si Aplicatii*, Presa Universitara Clujeana, 2009.
7. E. Dagotto, *Nanoscale Phase Separation and Colossal Magnetoresistance*, Springer-Verlag Berlin Heidelberg 2002.
8. J.M deTeresa, [New magnetic materials and their functions](http://esm.neel.cnrs.fr/2007-cluj) 2007, Cluj-Napoca, Romania. Summer School (<http://esm.neel.cnrs.fr/2007-cluj>)
9. UvA-VU Master Course: *Advanced Solid State Physics*
10. H. Ibach and H. Lüth: *Solid State Physics* 3rd edition (Springer-Verlag, Berlin, 2003) ISBN 3-540-43870-X
11. J. M. ZIMAN, *ELECTRONS AND PHONONS ,The Theory of Transport Phenomena in Solids*, UNIVERSITY OF CAMBRIDGE , OXFORD AT THE CLARENDON PRESS 1960

Introduction. Transport Processes

Nonequilibrium
steady state



Entropy (reservoir 1+reservoir 2+ System) **INCREASES**

Driving force=Temperature gradient



Transport of Energy

$$\text{Flux} = (\text{coefficient}) \times (\text{driving force})$$

Linear phenomenological law (if the force is not too large)

e.g. Ohm's law for the conduction of electricity

J_A = flux density of A = net quantity of A transported
across area in unit time

Net transport = the transport in one direction – the transport in opposite direction

$$\mathbf{J}_A = -D \cdot \text{grad } \rho_A$$

Table 14.2 Summary of phenomenological transport laws

Effect	Flux of particle property	Gradient	Coefficient	Law	Name of law	Approximate expression for coefficient
Diffusion	Number	$\frac{dn}{dz}$	Diffusivity D	$\mathbf{J}_n = -D \text{ grad } n$	Fick's law	$D = \frac{1}{3}\bar{c}l$
Viscosity	Transverse momentum	$M \frac{dv_x}{dz}$	Viscosity η	$\frac{F_x}{A} = J_p^x = -\eta \frac{dv_x}{dz}$	Newtonian viscosity	$\eta = \frac{1}{3}\rho\bar{c}l$
Thermal conductivity	Energy	$\frac{d\rho_u}{dz} = \hat{C}_v \frac{dT}{dz}$	Thermal conductivity K	$\mathbf{J}_u = -K \text{ grad } \tau$	Fourier's law	$K = \frac{1}{3}\hat{C}_v\bar{c}l$
Electrical conductivity	Charge	$-\frac{d\varphi}{dz} = E_z$	Conductivity σ	$\mathbf{J}_q = \sigma \mathbf{E}$	Ohm's law	$\sigma = \frac{nq^2l}{M\bar{c}}$

SYMBOLS: n = number of particles per unit volume
 \bar{c} = mean thermal speed = $\langle |v| \rangle$
 l = mean free path
 \hat{C}_v = heat capacity per unit volume
 ρ_u = thermal energy per unit volume
 F_x/A = shear force per unit area

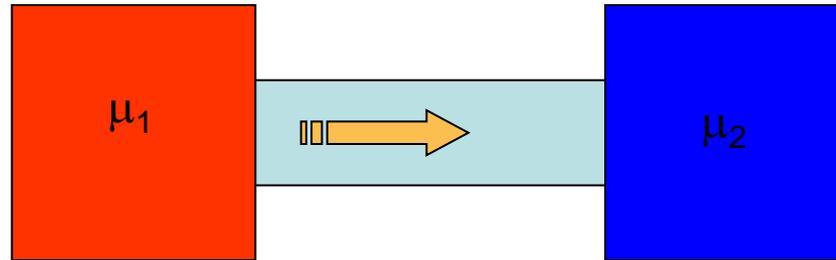
φ = electrostatic potential
 \mathbf{E} = electric field intensity
 q = electric charge
 M = mass of particle
 ρ = mass per unit volume
 \mathbf{p} = momentum

from Enss

Particle diffusion

from Kittel, Thermal Physics

$T = \text{constant}$



particle flow

Entropy (reservoir 1+reservoir 2+ System) **INCREASES**

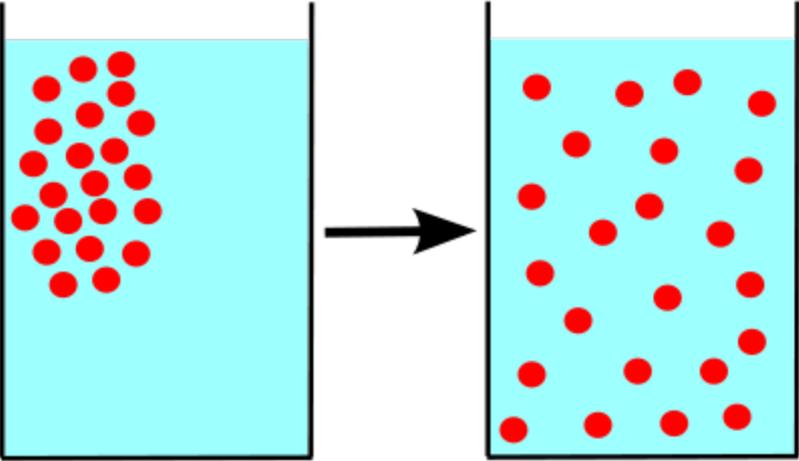
Presume: the difference of chemical potential is caused by a difference in particle concentration

J_n - the number of particles passing through a unit area in unit time

Fick's law:

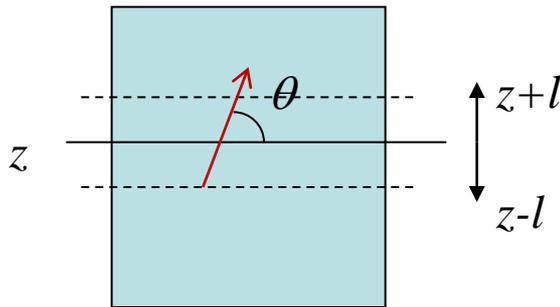
$$J_n = -D \cdot \text{grad } n$$

D - particle **diffusion constant** - **diffusivity**



l – the mean free path

At position z the particles come into a local equilibrium condition $\mu(z)$ and $n(z)$



$$J_n^z = \frac{1}{2} [n(z - l_z) - n(z + l_z)] \bar{c}_z = -\frac{dn}{dz} \bar{c}_z l_z$$

We express $c_z l_z$ in terms of $\bar{c} \cdot l$

$$l_z = l \cdot \cos \theta$$

$$\bar{c}_z = \bar{c} \cdot \cos \theta$$

The average is taken over the surface of a hemisphere

The element of surface area is $2\pi \cdot \sin \theta \cdot d\theta$

$$\langle \bar{c}_z l_z \rangle = \bar{c} l \frac{2\pi \int_0^{\pi/2} \cos^2 \theta \cdot \sin \theta \cdot d\theta}{2\pi} = \frac{1}{3} \bar{c} l$$

So that

$$J_n^z = -\frac{1}{3} \bar{c} l \frac{dn}{dz}$$



$$D = \frac{1}{3} \bar{c} l$$

Particle diffusion is the model for other transport problems

- Particle diffusion → transport of particles
- Thermal conductivity → transport of energy by particles
- Viscosity → transport of momentum by particles
- Electrical conductivity → transport of charge by particles

The linear transport coefficients that describes the processes are proportional to the particles diffusivity D

let

ρ_A – the concentration of the physical quantity A.

-the flux density of A in the z direction is:

$$J_A^z = \rho_A \langle v_z \rangle$$

$\langle v_z \rangle$ is the mean drift velocity of the particles in the z direction
(drift velocity is zero in thermal equilibrium)

If A (e.g. energy, momentum...) depends on the velocity of a molecule:

$$J_A^z = f_A \cdot \rho_A \langle v_z \rangle$$

f_A is a factor with magnitude of the order of unity → it depends on the velocity dependence of A and may be calculated (e.g. by using the Boltzmann transport equation)

By analogy with
Fick' law

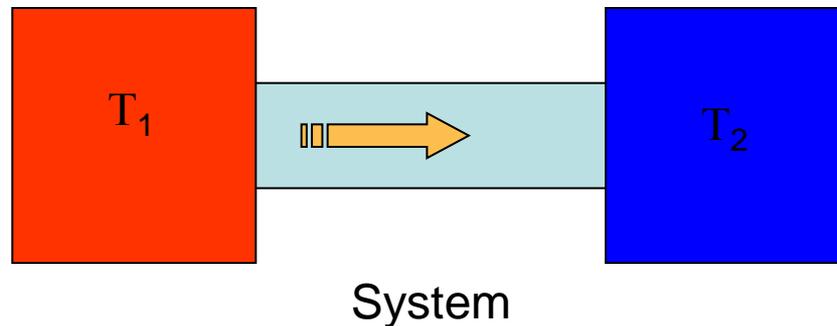
$$J_A = -D \cdot \text{grad } \rho_A$$

Thermal conductivity

Fourier law's

$$\mathbf{J}_A = -K \cdot \text{grad } T$$

Describes the energy flux density \mathbf{J}_n in terms of the **thermal conductivity** K and the temperature gradient



- This form assumes that there is a net transport of energy, but not particles
- Another term must be added if additional energy is transported by means of particle flow (as when electrons flow under the influence of an electric field.)

The energy flux density in the z direction is:

$$J_u^z \cong \rho_u \langle v_z \rangle$$

ρ_u – is the energy density

By analogy with the diffusion equation:

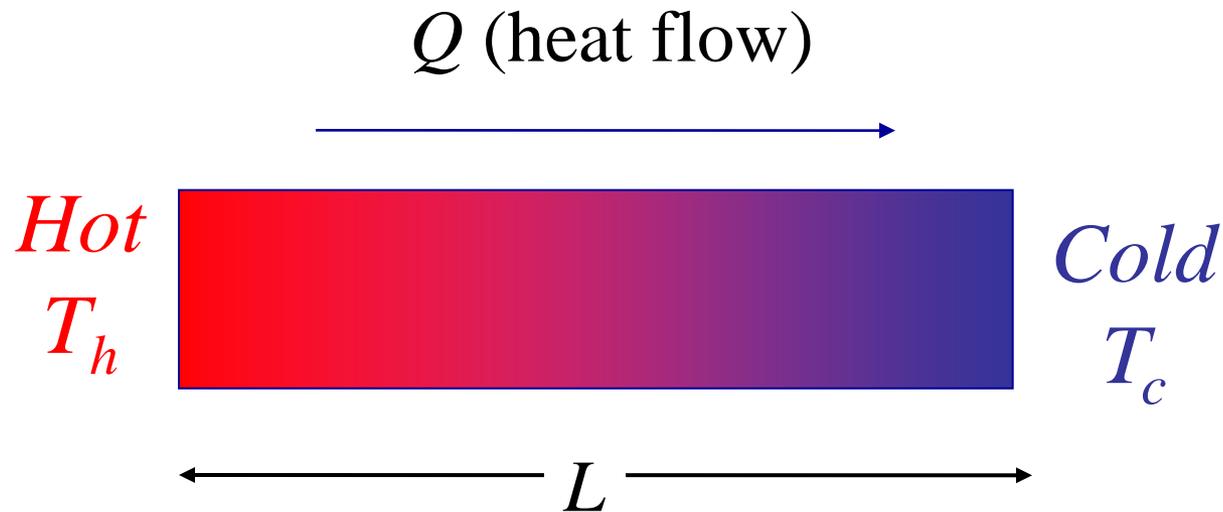
Diffusion of energy $-D \cdot d\rho_u / dx = -D(\partial\rho_u / \partial T)(dT / dx)$

$\partial\rho_u / \partial T$ Is the heat capacity per unit volume, C_V .

$$J_u = -D \cdot C_V \cdot \text{grad } T$$

$$K = DC_V = \frac{1}{3} C_V \cdot \bar{c} \cdot l$$

The thermal conductivity of a gas is independent of pressure **until very low pressure** when the mean free path becomes limited by the dimensions of the apparatus rather than by intermolecular collision.

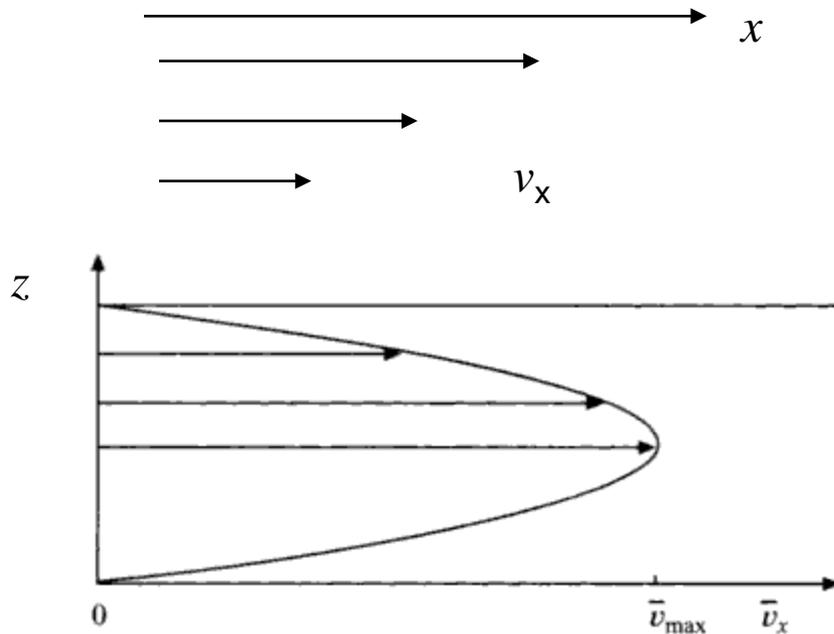


$$Q = kA \frac{T_h - T_c}{L} = kA \frac{dT}{dx}$$

↑
Thermal conductivity

Viscosity

Viscosity is a measure of the diffusion of momentum parallel to the flow velocity and transverse to the gradient of the flow velocity



Viscosity coefficient:

$$X_z = -\eta \frac{dv_x}{dz} = J_z(p_x)$$

from Enss

Figure 15.6 Velocity profile for slow fluid flow in a pipe. The velocity component \bar{v}_x is the velocity component along the pipe; the overbar denotes an average of molecular velocity components taken over a small volume (that is, small relative to the pipe radius R).

The particle flux density in the z direction: $J_n^z = n \langle v_z \rangle = -D dn/dz$

The transverse momentum density: nMv_x

Its flux density in the z direction: $(nMv_x) \langle v_z \rangle$

Kittel,
Thermal
physics

This flux density $-Dd(nMv_x)/dz$

$$(J_A = -D \cdot \text{grad } \rho_A)$$

Mass density: $\rho = nM$

$$J_z(p_x) = \rho v_x \langle v_x \rangle = -D \rho dv_x/dz = -\eta dv_x/dz$$

$$\eta = D\rho = \frac{1}{3} \rho \bar{c} l$$

CGS- poise

SI unit: Pa.s

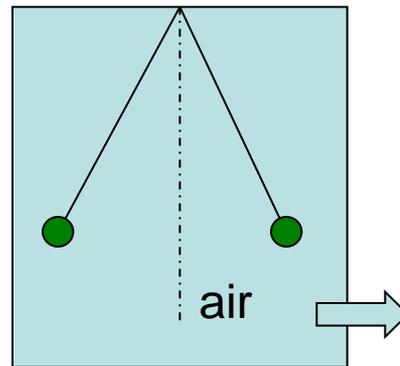
The mean free path: $l = 1 / \pi d^2 n$

Viscosity:

$$\eta = M\bar{c} / 3\pi d^2 \quad \text{independent of gas pressure}$$

The independence fails at **very high pressures** when the molecules are always in contact, or at **very low pressures** when the mean free path is longer than the dimension of the apparatus.

Robert Boyle 1660



$$D = \eta / \rho \quad \longrightarrow$$

$$K = \eta C_v / \rho$$

Kinematic viscosity

Generalized Forces

The transfer of entropy from one part of a system to another is a consequence of any transport process.

- We can relate the rate of change of entropy to flux density of particles and of energy

For $V = \text{const.}$
$$dS = \frac{1}{T} dU - \frac{\mu}{T} dN$$

The entropy current density :
$$\vec{J}_s = \frac{1}{T} \vec{J}_u - \frac{\mu}{T} \vec{J}_n \quad (*)$$

Entropy density \hat{S}

The net change of entropy density at a fixed position

$$\partial \hat{S} / \partial t \quad \longrightarrow$$

$$\partial \hat{S} / \partial t = g_s - \text{div } \vec{J}_s$$

Eq. of continuity

Rate of production of entropy

In a transfer process U and N are conserved

The equations of continuity:

$$\frac{\partial u}{\partial t} = -\text{div} \vec{J}_u$$

$$\frac{\partial n}{\partial t} = -\text{div} \vec{J}_n$$

Divergence of \vec{J}_S

$$\text{div} \vec{J}_S = \frac{1}{T} \text{div} \vec{J}_u + \vec{J}_u \cdot \text{grad}(1/T) - (\mu/T) \text{div} \vec{J}_n - \vec{J}_n \cdot \text{grad}(\mu/T)$$

(*)

$$\frac{\partial \hat{S}}{\partial t} = \frac{1}{T} \frac{\partial u}{\partial t} - \frac{\mu}{T} \frac{\partial n}{\partial t}$$

$$g_S = \vec{J}_u \cdot \text{grad}(1/T) + \vec{J}_n \cdot \text{grad}(-\mu/T)$$

$$g_S = \vec{J}_u \cdot \vec{F}_u + \vec{J}_n \cdot \vec{F}_n$$

Generalized forces

$$\vec{F}_u \equiv \text{grad}(1/T) \quad \vec{F}_n \equiv \text{grad}(-\mu/T)$$

Thermodynamics of irreversible processes:

$$\begin{aligned} \vec{J}_u &= L_{11}\vec{F}_u + L_{12}\vec{F}_n \\ \vec{J}_n &= L_{21}\vec{F}_u + L_{22}\vec{F}_n \end{aligned}$$

Coupled effects

Onsager relation:

$$L_{ij} \begin{pmatrix} \vec{B} \\ \vec{B} \end{pmatrix} = L_{ji} \begin{pmatrix} \vec{B} \\ -\vec{B} \end{pmatrix}$$

In magnetic field

Advanced Treatment: Boltzmann Transport Equation

We work in the 6 D (six-dimensional space of Cartesian coordinates r and v).

The classical distribution function:

$$f(\vec{r}, \vec{v}) d\vec{r} d\vec{v} = \text{number of particles in } d\vec{r} d\vec{v}$$

The effect of time displacement dt on the distribution function:

$$f(t + dt, \vec{r} + d\vec{r}, \vec{v} + d\vec{v}) = f(t, \vec{r}, \vec{v})$$

In the absence of collisions

With collisions:

$$f(t + dt, \vec{r} + d\vec{r}, \vec{v} + d\vec{v}) - f(t, \vec{r}, \vec{v}) = dt \left(\frac{\partial f}{\partial t} \right)_{\text{collisions}}$$

With a series development:

$$dt \left(\frac{\partial f}{\partial t} \right) + d\vec{r} \cdot \text{grad}_{\vec{r}} f + d\vec{v} \cdot \text{grad}_{\vec{v}} f = dt \left(\frac{\partial f}{\partial t} \right)_{\text{collisions}}$$

$$\vec{a} \equiv \frac{d\vec{v}}{dt}$$

$$\frac{\partial f}{\partial t} + \vec{v} \cdot \text{grad}_{\vec{r}} f + \vec{a} \cdot \text{grad}_{\vec{v}} f = \left(\frac{\partial f}{\partial t} \right)_{\text{collisions}}$$

Boltzmann transport equation

Relaxation time approximation:

This is based on the assumption that a nonequilibrium distribution $f(t, \vec{r}, \vec{v})$ gradually returns to its equilibrium value within a characteristic time, the relaxation time $\tau_c(\vec{r}, \vec{v})$, by scattering of particles with the velocity \vec{v} into states \vec{v}' , and vice versa.

$$\left(\frac{\partial f}{\partial t} \right)_{\text{collisions}} = -(f - f_0) / \tau_c$$

Suppose that a nonequilibrium distribution of velocities is set up by external forces which are suddenly removed.

The decay of the distribution towards equilibrium is then obtained:

$$\frac{\partial(f - f_0)}{\partial t} = -(f - f_0) / \tau_c$$

$$\partial f_0 / \partial t = 0$$

$$(f - f_0)_t = (f - f_0)_{t=0} \exp(-t / \tau_c)$$

Generally

$$\tau_c = \tau_c(\vec{r}, \vec{v})$$

$$\partial f / \partial t + \vec{v} \cdot \text{grad}_{\vec{r}} f + \vec{a} \cdot \text{grad}_{\vec{v}} f = \frac{f - f_0}{\tau_c}$$

In the steady state:
by definition $\partial f / \partial t = 0$

Particle Diffusion

Consider an isothermal system with a gradient of particle concentration

The steady-state Boltzmann transport equation in the relaxation time approximation:

$$v_x df/dx = -(f - f_0)/\tau_c$$

First order approximation

$$f_1 \cong f_0 - v_x \tau_c df_0/dx \quad \frac{df}{dx} \rightarrow \frac{df_0}{dx}$$

Second order approximation

$$f_2 \cong f_0 - v_x \tau_c df_1/dx = f_0 - v_x \tau_c df_0/dx + v_x^2 \tau_c^2 d^2 f_0/dx^2$$

The iteration is necessary for the treatment of nonlinear effects

Classical Distribution

$$f_0 = \exp[(\mu - \varepsilon) / k_B T]$$

$$df_0 / dx = (df_0 / d\mu)(d\mu / dx) = (f_0 / k_B T)(d\mu / dx)$$

The first order solution for the nonequilibrium distribution becomes:

$$f = f_0 - (v_x \tau_c f_0 / k_B T)(d\mu / dx)$$

The particle flux density in the x direction:

$$J_n^x = \int v_x f D(\varepsilon) d\varepsilon$$

The density of orbitals per unit volume per unit energy range:

$$D(\varepsilon) = \frac{1}{4\pi^2} \left(\frac{2M}{\hbar^2} \right)^{3/2} \varepsilon^{1/2}$$

Presume τ_c constant, independent of velocity

$$\int f_0 D(\varepsilon) d\varepsilon = n$$

finally,

$$J_n^x = -(n\tau_c / M)(d\mu / dx) = -(k_B T \tau_c / M)(dn / dx)$$



because $\mu = k_B T \log n + \text{const.}$

Diffusivity:

$$D = \tau_c k_B T / M = \frac{1}{3} \langle v^2 \rangle \tau_c$$

If we presume

$$\tau_c = \frac{l}{v} \quad \longrightarrow$$

$$D = \frac{1}{3} l \bar{c}$$

Fermi-Dirac distribution

$$f_0 = \frac{1}{\exp[(\varepsilon - \mu)/k_B T] + 1}$$

$$df_0/d\mu \cong \delta(\varepsilon - \mu) \quad \int_{-\infty}^{+\infty} F(\varepsilon) \delta(\varepsilon - \mu) d\varepsilon = F(\mu)$$

$$df_0/dx = \delta(\varepsilon - \mu) d\mu/dx$$

The particle flux density

$$J_n^x = \int v_x f D(\varepsilon) d\varepsilon = -(d\mu/dx) \tau_c \int v_x^2 \delta(\varepsilon - \mu) D(\varepsilon) d\varepsilon$$

We know

$$D(\mu) = 3n/2\varepsilon_F \quad \mu = (\hbar^2/2m)(3\pi^2 n)^{2/3}$$

$$J_n^x = -(2\tau_c/3m)\varepsilon_F dn/dx = -\frac{1}{3}v_F^2 \tau_c dn/dx$$

$$\frac{1}{3}v_F^2 \left(\frac{3n}{2\varepsilon_F}\right) = n/m$$

$$D = \frac{1}{3}v_F^2 \tau_c$$

Electrical conductivity

We multiply the particle **flux density** by the particle charge q

$$d\mu / dx \implies qd\phi / dx = -qE_x$$

$$J_q^x = (n\tau_c q / m)(d\mu / dx) = (n\tau_c q^2 / m)E = \sigma E$$

Electrical conductivity

$$\sigma = nq^2\tau_c / m$$

- For a classical gas (Drude)
- For the Fermi-Dirac distribution (Sommerfeld)

We will discuss this in more detail, later.